AOSC201: Weather and Climate Lab

Week 9: Weather and Air Quality

Section 103/105

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Tropospheric Ozone

- 2015 National Ambient Air Quality Standards (NAAQS): 70 ppb at an 8-hr avg
 - 2008: 75 ppb for 8-hr avg
 - 1997: 80 ppb for 8-hr avg
 - 1979: 120 ppb for 1-hr avg
- Produced by reactions with sunlight, natural gases, and incomplete combustion reaction products
- Peaks mid-afternoon
- Destroyed or dispersed by nitric oxide, titration, wind transport, etc.
 Ozone production and destruction with no

hydrocarbons

Afternoon Midday Morning

CO O₃ NO₂ CO NO

VOCS

CO VOCS

Photochemical smog diurnal evolution

$$NO_2 + hv \rightarrow NO + O$$

$$O + O_2 + M \rightarrow O_3 + M$$

$$NO + O_3 \rightarrow NO_2 + O_2$$

Tropospheric Ozone

- Known to create a great deal of respiratory issues in plants and humans alike
- O₃ diurnal cycle: surface heating begins and interacts with exhaust products to produce ozone towards the afternoon
 - Commonly referred to as photochemical smog



Photochemical smog diurnal evolution

Ozone production and destruction with no hydrocarbons

$$NO_2 + hv \rightarrow NO + O$$

 $O + O_2 + M \rightarrow O_3 + M$
 $NO + O_3 \rightarrow NO_2 + O_2$

National Ambient Air Quality Standards (NAAQS)

- Primary: provide public health protection, including protecting the health of "sensitive" populations such as asthmatics, children, and the elderly
- Secondary: provide public welfare protection, including protection against decreased visibility and damage to animals, crops, vegetation, and buildings
- 6 criteria pollutants: ozone, lead, particulate matter (PM10, PM2.5), sulfur dioxide, carbon monoxide, nitrogen dioxide

Current ozone standards (NAAQS)

| Pollutant [links to historical tables of NAAQS reviews] | Primary/ Secondary | Averaging Time | Level | Form |
|---|-----------------------------|-------------------|--------------------------|---|
| <u>Ozone (O₃)</u> | primary and secondary | 8 hours | 0.070 ppm ⁽³⁾ | Annual fourth-highest daily maximum 8-hour concentration, averaged over 3 years |

Air Quality Index (AQI)

- EPA's index for reporting air quality
- The higher the AQI value, the greater the level of air pollution and the greater the health concern
- BP: break points (on your lab manual, pg 53)
- I: index for pollutant p
- Cp: rounded concentration of pollutant

$$I_{p} = \frac{I_{Hi} - I_{Lo}}{BP_{Hi} - BP_{Lo}} (C_{p} - BP_{Lo}) + I_{Lo}$$

| Air Quality Index Levels of Health Concern | Numerical Value | Meaning | |
|--|--------------------|--|--|
| Good | 0 to 50 | Air quality is considered satisfactory, and air pollution poses little or no risk | |
| Moderate | 51 to 100 | Air quality is acceptable; however, for some pollutants there may be a moderate health concern for a very small number of people who are unusually sensitive to air pollution. | |
| Unhealthy for Sensitive Groups | 101 to 150 | Members of sensitive groups may experience health effects. The general public is not likely to be affected. | |
| Unhealthy | 151 to 200 | Everyone may begin to experience health effects; members of sensitive groups may experience more serious health effects. | |
| Very Unhealthy | 201 to 300 | Health alert: everyone may experience more serious health effects | |
| Hazardous | 301 to 500 | Health warnings of emergency conditions. The entire population is more likely to be affected. | |

Air Quality Index (AQI) Sample Calculation

For ozone concentration of 30 ppb:

$$x_{ppm} = \frac{x_{ppb}}{1000}$$
$$x_{ppm} = \frac{30 \text{ ppb}}{1000}$$

 $x_{ppm} = 0.030 ppm$

| Ozone Breakpoints (ppm) | AQI |
|-------------------------|--------|
| 0.000 - 0.064 | 0 - 50 |

$$I_{p} = \frac{I_{Hi} - I_{Lo}}{BP_{Hi} - BP_{Lo}} (C_{p} - BP_{Lo}) + I_{Lo}$$

$$I_{p} = \frac{50 - 0}{0.064 - 0.000} (0.030 \text{ ppm} - 0.000) + 0$$

$$I_p = 23$$
 — AQI: Good, air quality is considered satisfactory, and air

AQI: Good, air quality is pollution poses little to no risk

HINT: look for your ozone concentration in ppb, convert to ppm, then find what range your concentration fits in in the breakpoints column.

Calculating 8-hour O₃ averages

$$8hr\ avg = \frac{Cp_{hr\,1} + Cp_{hr\,2} + Cp_{hr\,3} + Cp_{hr\,4} + Cp_{hr\,5} + Cp_{hr\,6} + Cp_{hr\,7} + Cp_{hr\,8}}{8}$$

- Convert to ppm (conversion on previous slide) for your final answer!!!!
- Do 12am to 4pm
 - You don't need to calculate averages from 5pm to 11pm (see extra directions)
- LAB FREEBIE using chart given in manual/extra directions at 12am:

$$12am_{8hr} = \frac{50\ ppb + 46\ ppb + 40\ ppb + 48\ ppb + 22\ ppb + 20\ ppb + 25\ ppb + 44\ ppb}{8}$$

$$12am_{8hr} = 36.875 \ ppb \quad \rightarrow \quad \frac{36.875 \ ppb}{1000} \approx 0.036875 \ ppm$$

For #1, leave the concentration in ppb in the table!